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# CANADIAN PATENT

⑤④ PHENOXY-ALKYL-CARBOXYLIC ACID DERIVATIVES AND  
THE PREPARATION THEREOF

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The invention concerns new derivatives of phenoxy-alkyl-carboxylic acids and their method of preparation, characterised in that para-hydroxy-benzaldehyde, a para-hydroxy-phenone, or p-hydroxybenzoic acid is condensed with chloro-acetic acid or one of the superior homologues thereof, the acid or the diacid obtained being possibly converted afterwards.

The acid obtained can be converted into an ester according to the classic sulphuric esterification method in the presence of a chosen alcohol.

The ester obtained can be also converted into an amide by boiling to reflux in the presence of chosen aromatic or aliphatic amide or its derivatives.

The amide can also be obtained by means of an acid chloride according to the usual methods, from the corresponding acid.

The aldehyde or cetone function is converted into an axime by heat treatment in the presence of hydroxylamine chlorohydrate.

The acid obtained is possibly converted into its halogenated para- $\omega$  derivative by the addition of a halogen to a solution of the said acid in acetic acid in the presence of acetic anhydride.

The para- $\omega$  halogen derivatives of the acid can also be converted into an ester, amide or oxime.

The ester or oxime can also be converted into phenoxy-alkyl, carbohydroxamic acid by bringing them to reflux in ethyl alcohol containing hydroxylamine and sodium.



The dehydration of the para-formyl-oxime-phenox-alkyl carboxylic acid esters by boiling to reflux in the presence of acetic anhydride gives new chemical bodies as the esters of para-cyano-phenox-alkyl-carboxylic acids. The corresponding amides are  
 5 obtained by boiling the said esters in the chosen amine.

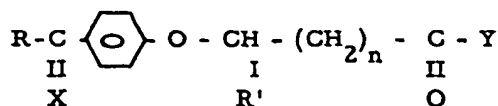
The esters of para-cyano-phenox-alkyl-carboxylic acids are converted into para-carboxamide-phenox-alkyl-carboxylic acids by oxygenated water in a warm medium of alcalin. These acids can be, in their turn, esterified then converted into an amide.

10 According to another characteristic of the invention the para-carboxy-phenox-alkyl-carboxylic acids diacids obtained according to the general process can then possibly be submitted to an esterification or an amidification or can be converted into a corresponding hydroxamic acid.

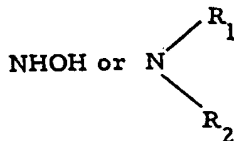
15 The new compounds obtained by these processes are particularly remarkable for their therapeutic applications in neutrope, anti-inflammation, normo-lipemiant medicines, for example.

The formulae of the chemical compounds according to the invention are characterised by the presence on a phenyl  
 20 ring of a cetone aldehyde group, convertible into an oxime, acid, ester, amide, hydroxamic or nitrile acid in para position in relation to the oxy-alkyl group, or by the presence of a halogen in  $\omega$  position of the chain in para position in relation to the oxy-alkyl group, and by the presence of an oxy-alkyl chain of 2, 3 or 4 carbon atoms supporting an acid, ester,  
 25 amide, or carbohydroxamic function.

The derivatives according to the invention comply to the general formula :

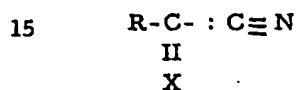


where R is -H or  $\text{NH}_2$ ,  $-\text{CH}_3$ ;  $-\text{CH}_2-\text{CH}_3$ ;  $-\text{CH}_2-\text{CH}_2-\text{CH}_3$ ;  $-\text{C}_6\text{H}_5$  or the  $\omega$   
 5 halogen derivatives of the preceding groups,  $-\text{OH}$ ;  $-\text{OCH}_3$ ;  $\text{OC}_2\text{H}_5$ ;



(where  $\text{R}_1$  and  $\text{R}_2$  can be linear radicals such as  $-\text{CH}_3$  or  $-\text{CH}_2\text{CH}_3$  or a  
 10 cycle such as piperidine, methyl 2 piperidine, piperazine, morpholine,  
 pyrrolidine, methyl 4 piperidine N-phenyl piperazine, N-p methoxy-phenyl-  
 piperazine, N-methyl 4 piperazine, N-p chlorophenyl piperazine or hex-  
 amethyleimine or ethylamino ethylamine).

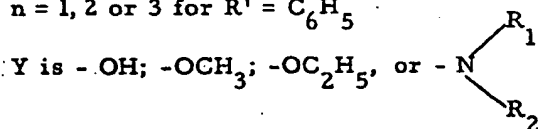
X is = O or = N-O-H



$\text{R}'$  is -H, or  $-\text{C}_6\text{H}_5$

$n = 0, 1, 2$  or  $3$  normo or iso for  $\text{R}' = \text{H}$

20  $n = 1, 2$  or  $3$  for  $\text{R}' = \text{C}_6\text{H}_5$



(where  $\text{R}_1$  and  $\text{R}_2$  can be radical linears such as  $\text{CH}_3$  or  $\text{CH}_2\text{CH}_3$  or rep-  
 resented by a cycle such as piperidine methyl 2 piperidine, piperazine,  
 25 morpholine, pyrrolidine, methyl 4 piperidine, N-phenyl piperazine, N-p  
 methoxy-phenyl piperazine, N-methyl 4 piperazine, N-p chlorophenyl  
 piperazine or hexamethyleimine, or ethyl amino ethyl amine.)

The invention will be better understood by referring  
 to the following description detailing the preparation of some of the new

chemical compounds by way of example.

A : Synthesis of p-ceto-alkyl-carboxylic or p-formyl-alkyl-carboxylic acids and their derivatives.

I - Synthesis of acids

5

0, 6 mole of Na OH

0, 3 mole of p-hydroxy-benzaldehyde or of p-hydroxy-acetophenone or of its superior homologues propiophenone and butyphenone.

10

0, 3 mole of chloroacetic, chloropropionic or chlorobutyric-acids are introduced into a balloon-flask containing 600 ml. of water.

15

The solution is brought to reflux boiling for 8 hours. After cooling it is acidified with  $\text{pH}_3$ , by 12N HCl. The acid precipitates and is filtered. The product obtained is redissolved in ether and extracted by an acidified aqueous solution. The acid is recrystallised in water by cooling to the ambient temperature.

The average yield of this operation is in the order of 65%.

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By this method of synthesis five types of products deriving from para-formyl-phenoxy-alkyl-carboxylic acid, from para acetyl-phenoxy-alkylcarboxylic acid, from para propionyl-phenoxy-alkyl-carboxylic acid, from para-butyryl-phenoxy-alkylcarboxylic acid and from para-benzoyl-phenoxy-alkyl- carboxylic acid can be obtained.

The alkyl derivatives corresponding to each of the five series can be an acetic propionic chain (normo or iso).

25

In order to obtain the homologues of these five series for which the alkyl link is an iso-butyric link, a slightly different method must be employed, which consists of : in a three-column balloon flask of

1,000 cm<sup>3</sup>, provided with mechanical stirrer and a reflux refrigerant surmounted by a Ca Cl<sub>2</sub> column is introduced 0,1 mole of para-hydroxyphenylacetone dissolved in 100 ml of anhydrous acetone.

Under agitation 0,5 mole of Na OH in pulverised pastiles is added. It is brought up to reflux by heating in a water bath maintaining the agitation. Sodium phenate precipitates after about 15 to 20 minutes.

10 ml of anhydrous CHCl<sub>3</sub>, diluted in 20 ml of anhydrous acetone is introduced by a bromine flask.

Reflux heating is then continued for four hours.

300 ml of water is then added and the evaporation of the acetone is obtained in a vacuum. The aqueous solution is then acidified to pH3 with 12N HCl then extracted twice over by 200 ml of ether.

The ethereal solution is extracted in two successive times by 150 ml of saturated solution of sodium carbonate acid.

By acidification of the carbonated solutions to pH3 by 12N HCl an oil is formed which is extracted by 400 ml of ether. The ethereal solution is poured off, dried by Na<sub>2</sub>SO<sub>4</sub>, then evaporated in a vacuum.

The phenoxy-isobutyric acid takes the form of a mass. It is recrystallised in a mixture of alcohol and water.

Then phenoxy-isobutyric acids obtained according to this method are synthesised with a yield of 50%.

By way of example, the following products exhibit the following particular properties :

a) Para-butyryl-phenoxy-isobutyric acid

Fusion point : 88°C

Soluble in ether, alcohol and acetone

Insoluble in water

b) Para-benzoyl phenoxy-isobutyric acid :

Fusion point  $130^{\circ}\text{C}$

Soluble in ether, alcohol and acetone

5

Insoluble in water

## II - Esterification

10 g of an acid obtained in I are dissolved in 150 ml of methanol, ethanol or propanol.

150 ml of anhydrous benzine

10

1 ml of  $36\text{N H}_2\text{SO}_4$  are gradually added to a flask.

After boiling to reflux for two hours, the azeotrope benzine-alcohol is distilled until all the benzine is eliminated.

The alcoholic solution is next concentrated in a vacuum. The oil obtained is recovered by 200 ml ethyl ether, the solution is washed in water then dried on  $\text{Na}_2\text{SO}_4$ . After concentration of the ethereal solution in a vacuum the ester is obtained in the form of a yellow oil or in crystallised form according to the case.

15

### Examples

a) The ethylic ester of the p -acetyl-phenoxy-iso butyric acid is obtained after dissolving the acid in absolute ethanol, the addition of benzine and  $\text{H}_2\text{SO}_4$ , boiling to reflux then purification; it is a liquid soluble in alcohol, ether and chloroform, and in water; its boiling point is  $120^{\circ}\text{C}$  at 0,03 mm Hg. The yield of the preparation is in the neighbourhood of 85%.

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b) The ethyl esters of para butyl phenoxy iso butyric acid is obtained in the same conditions from p butyl-phenoxy-acetic acid, with a yield of 70%. It is a liquid soluble in alcohol and ether,

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insoluble in water and its boiling point is  $144^{\circ}\text{C}$  at 0,05 mm Hg.

### III - Amidification

#### 1) From esters

8 g of an ester is obtained and dissolved in about  
 5 25 ml of an amine previously dried on potash. The solution is refluxed for  
 3 hours. The amine generally crystallises by simple cooling or by the  
 addition of a slight quantity of water. The complete precipitation is ob-  
 tained by the slow addition of 200 to 300 ml of water. Purification is  
 carried out by recrystallation in a mixture of alcohol and water.

#### 10 2) From acid chlorates

It is possible to treat the acid chlorates obtained  
 according to the usual methods, in the same conditions as that above in  
 III 1) in the presence of amine in order to obtain the required amide.

The following are given by way of example :

15 a) Morpholine amide of p-formyl phenoxy acetic  
 acid is obtained according to 1 or 2 above. The yield is 65% and the pro-  
 duct is soluble in alcohol and insoluble in water; the fusion point for the  
 amide is  $116^{\circ}\text{C}$ .

b) The amide of p chloro-phenyl 4 piperazine and of  
 20 the p formyl phenoxyacetic acid is obtained by methods described in 2  
 above, with a yield of 45%; its fusion point is  $120^{\circ}\text{C}$ .

c) The amide of p chlorophenyl 4 piperazine and of  
 p acetyl phenoxy acetic acid is obtained according to 1 above with a yield  
 of about 40%. Its fusion point is  $115^{\circ}\text{C}$ ; it is soluble in alcohol, slightly  
 25 soluble in ether, and insoluble in water petroleum ether and hexane.

d) p (acetyl-phenoxy acetyl) 1 morpholine is obtain-  
 ed according to 1 above. It is soluble in alcohol, insoluble in water, and



petroleum ether. Its fusion point is  $112^{\circ}\text{C}$ , with a yield of 60%.

5 e) (p acetyl phenoxy acetyl) 1-methyl 4 piperidine is obtained according to 1 above with a yield in the order of 40% its fusion point is  $60^{\circ}\text{C}$ ; it is soluble in alcohol and ether, insoluble in water and petroleum ether.

f) (p acetyl phenoxy acetyl) 1-hexamethylemimine is obtained according to 1 above with a yield of 50%; it is soluble in alcohol and ether insoluble in water. Its fusion point is  $78^{\circ}\text{C}$ .

10 g) N (p acetyl phenoxy acetyl) diethyl amino ethyl amine is obtained according to 1 above, with a yield of 35%. It is insoluble in water and petroleum ether, soluble in alcohol and ether, and its fusion point is  $75^{\circ}\text{C}$ .

15 h) Piperidine amide p formyl phenoxy acetic acid is obtained by the treatment indicated above in 1 or 2. A crystallised product is obtained of which the fusion point is  $96^{\circ}\text{C}$ , it is soluble in ether, alcohol and most organic solvents and insoluble in petroleum ether and water. The yield of the preparation is about 60%.

20 i) Piperidine amide of p acetyl phenoxy acetic acid is obtained in the conditions described in 1 above. A crystallised product is obtained the fusion point of which is  $97^{\circ}\text{C}$  with a yield of about 60%. This product is soluble in alcohol and most organic solvents, but insoluble in water and petroleum ether.

The amides complying to the general formula for which,  $\text{R}' = \text{C}_6\text{H}_5$  are also obtained by applying the processes 1 and 2 above.

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#### IV - Production of oximes

0,1 mole of an amide obtained as in III is made into a solution with 500 ml of absolute ethanol. 7g hydroxylamine and chlorhy-

drate and 5, 8 g of soda are added to the solution which is then refluxed for three hours. After the addition of 100 ml of water, the oxime precipitates in the hydro-alcoholic solution, by concentration in a vacuum. After filtration, the oxime is purified by recrystallisation in a mixture of alcohol and water.

Examples:

a) The oxime of the piperidine amide of p formyl phenoxy-acetic acid is obtained in the above conditions. The crystallised oxime is obtained with a yield of about 70% : its fusion point is 135°C. It is soluble in alcohol but insoluble in water.

b) (p formyl oxime phenoxy acetyl) 1 morpholine is obtained in the above conditions. It is soluble in alcohol, insoluble in water, its fusion point is 169°C and the yield is 60%.

c) p acetyloxime of the phenoxy acetic acid is obtained in the previous conditions for this variant after the PH alkaline obtained by the piperidine used in equimolecular proportions with the chlorhydrate of hydroxylamine. The product is obtained with a yield of 55%. It is soluble in alcohol and bicarbonated water, insoluble in water.

d) p ecetyl oxime of phenoxy ethyl acetate is obtained in the same conditions as for (b) from the p acetyl phenoxy acetic ester with<sup>a</sup> yield of 65%. It is soluble in alcohol, insoluble in water. Its fusion point is 103°C.

e) p acetyl oxime of the phenoxy-acetyl amido piperidine is obtained by condensation of the corresponding amide with chlorhydrate of hydroxylamine in alcohol in the presence of soda, boiling to reflux, then purification. The yield is 70%. The product is soluble in alcohol and insoluble in water; its fusion point is 168°C.

f) (p acetyl oxime phenoxy acetyl) 1 morpholine is obtained according to the same process with a yield of 70%. The product is soluble in alcohol, insoluble in water. Its fusion point is  $145^{\circ}\text{C}$ .

5 g) (p acetyl oxime phenoxy acetyl) 1 methyl 4 piperidine is obtained with a yield of 60%. The product is soluble in alcohol, insoluble in water; its fusion point is  $166^{\circ}\text{C}$ .

h) (p acetyl oxime phenoxy acetyl) 1 chlorophenyl 4 piperazine is obtained with a yield of 60%. The product is soluble in alcohol, very slightly soluble in ether, insoluble in water. Its fusion point is  $194^{\circ}\text{C}$ .

i) (p acetyl oxime phenoxy acetyl) 1 hexamethylmimine is obtained with a yield of 60%. It is soluble in alcohol, insoluble in water. Its fusion point is  $134^{\circ}\text{C}$ .

15 j) N (p acetyl oxime phenoxy acetyl) diethyl amino ethyl amine is obtained in the above conditions with a yield of about 50%. It is soluble in alcohol, insoluble in water; its fusion point is  $130^{\circ}\text{C}$ .

The oximes derived from compounds complying with the general formula for which  $\text{R}' = \text{C}_6\text{H}_5$  are also obtained according to the above process.

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#### V - Halogen derivatives

Another process according to the invention consists of the conversion of the phenoxy alkyl carboxylic acids into their halogen derivatives, the formula for which remains consistent with the for general formula but/which R signifies  $\text{ACH}_2 - (\text{CH}_2)_n$ , A representing a halogen in  $n' = 0, 1, 2, 3$  normo or iso positions.

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The fixing of the halogen on the chain, bromine for example, is obtained by the action of the bromine on phenoxy alkyl

carboxylic acid diluted in acetic acid. The bromine is added to the solution drop by drop in the presence of acetic anhydride, the precipitate obtained being washed and recrystallised after being dissolved in boiling water.

Such halogenous  $\omega$ acids complying to the general  
5 formula can be converted into their esters, amides, and oximes by application of the methods described in II, III and IV.

Examples :

1 - p( $\omega$ ) bromo acetyl phenoxy acetic acid is obtained  
ed by bromination as described above.

10 The product is obtained with a yield of about 70%.  
Its fusion point is  $183^{\circ}\text{C}$ . It is soluble in alcohol and in warm water,  
insoluble in cold water.

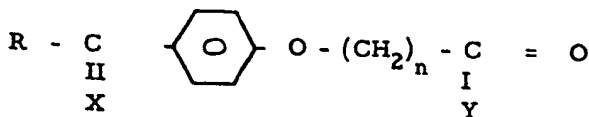
2 - p( $\omega$ ) bromo phenoxy ethyl acetate is obtained by  
esterification of the preceding acid conforming to the conditions described  
15 in II with a yield of about 90%. The product is soluble in alcohol, benzene  
and insoluble in water. Its fusion point is  $80^{\circ}\text{C}$ .

3 - (p( $\omega$ ) bromo acetyl phenoxy acetyl) 1 -piperidine  
is obtained by condensation of the previous ester and of the piperidine in  
the conditions described in III (1) = with a yield of about 50%. The product  
20 is soluble in warm alcohol, insoluble in water, and its fusion point is  $96^{\circ}\text{C}$ .

4 - (p( $\omega$ ) bromo acetyl oxime phenoxy acetyl) 1  
piperidine is obtained by condensation of the previous product with hydroxy-  
amine in the presence of soda, with a yield in the order of 40%. It is  
soluble in the usual solvents, its point of fusion is  $220^{\circ}\text{C}$ .

25 VI - Phenoxy alkyl carbohydroxamic acids and their  
derivatives

Phenoxy alkyl carboxylic acids and their derivatives  
of the general formula :



can be converted in order to obtain the phenoxy alkyl carbohydroxamic acids and their derivatives where Y is NHOH.

The general method consists of dissolving the corresponding esters or their p alkyl oxime derivatives in ethyl alcohol containing hydroxylamine and sodium, to be brought to reflux, then to precipitate the acid bringing the solution obtained having previously had the alcohol removed in an acid medium.

Examples :

a) p formyl-oxime phenoxy-aceto hydroxamic acid is obtained by dissolving in 500 ml of alcohol :

2-atom grams of sodium, 1 mole of hydroxylamine chlorhydrate, and 1 mole of p formyl oxime phenoxy ethyl acetate obtained according to (IV) or 3 atom grams of sodium, 2 moles of hydroxylamine chlorhydrate and of p formyl phenoxy ethyl acetate obtained according to (II).

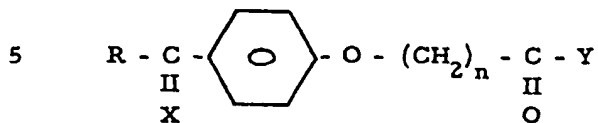
After cooling to reflux the NaCl formed is eliminated by filtration : 200 ml of water is added and the alcohol evaporated in a vacuum. The aqueous solution is acidified to pH3. The precipitate obtained is filtered and recrystallised in alcohol.

The product is obtained with a yield of about 50%. It is soluble in alcohol, insoluble in water, its fusion point is 198°C.

b) p acetyl oxime phenoxy aceto hydroxamic acid is obtained according to the previous procedure from the oxime phenoxy acetate of ethyl or the p acetyl phenoxy ethyl acetate with a yield of 55%.

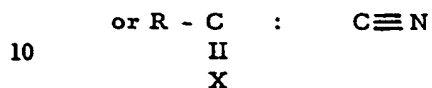
VII - Para cyano alkyl carboxylic acids

Another process according to the invention consists of transforming the p formyl phenoxy alkyl carboxylic acids of such type that their general formulae would be :



with X representing N-OH

R representing NH<sub>2</sub>



The process consists of dehydrating the esters of p formyl oxime phenoxy alkyl carboxylic acids by acetic anhydride during boiling to reflux. The esters of cyano phenoxy alkyl carboxylic acids are thus obtained. The corresponding amides can be obtained by boiling to reflux the preceding body dissolved in the corresponding amine or by dehydration of the (p-formyl-oxime-phenoxy-alkyl) 1 amido by acetic anhydride.

The p cyano phenoxy alkyl carboxylic acids can be converted into p carboxamide phenoxy alkyl carboxylic acids by oxygenated water in a warm alkaline medium. These acids can be esterified then converted into amides according to the methods described in II and III.

The (p carboxamide oxime phenoxy alkyl) 1 amidos are obtained directly from the p-carboxamide phenoxy alkyl carboxylic acids, by amidification according to III.

Examples :

a) p cyano phenoxy ethyl acetate is obtained by the following successive operations :

- Conversion of the p formyl phenoxy ethyl acetate into its oxime by condensation of hydroxylamine chlorhydrate in the presence of pyridine anhyde in equimolecular proportions by boiling to reflux.

- After recrystallisation of the previous body, 15 g of this are brought to reflux in 100 ml. of acetic anhydride. It is next hydrolyzed by 100 ml of warm water; after cooling, it is alkalized by 100 g of sodium bicarbonate and the p-cyano derivative is precipitated and recrystallised in the minimum of alcohol.

Thus, 9.5 g of the product are obtained, its fusion point being 57°C.

b) the p carboxamide phenoxy acetic acid is obtained by the action of 15 ml of  $H_2O_2$  110 volumes on 8 g of the previous bodies warmed in 100 ml of water containing 3 g of potash. After cooling, the acid is precipitated by the addition of 12N HCl up to PH3. 7 g of acid are obtained; its fusion point is 250°C.

c) p-carboxamide phenoxy ethyl acetate is obtained by boiling to reflux 8 g of the previous acid in 100 ml of absolute alcohol and 100 ml of benzene. After distillation of the azeotrope it is concentrated in a vacuum, the ester precipitates and it is next recrystallised in the minimum of alcohol to 95% . 5g of ester are obtained; its fusion point is 143°C.

d) (p carboxamide phenoxy acetyl) 1 piperidine is obtained from the preceding ester according to the method described in III (1) with a yield in the order of 55%. Its fusion point is 168°C.

e) (p cyano phenoxy acetyl) 1 piperidine is obtained either by dehydration from (p formyl oxime phenoxy acetyl) 1 piperidine by warm acetic anhydride or by treatment with p cyano phenoxy ethyl acetate according to the process described in III (1). The yield is 90%

The product has a fusion point of  $112^{\circ}\text{C}$ . It is soluble in alcohol and in ether, insoluble in water.

f) (p carboxamide oxime phenoxy acetyl) 1 piperidine is obtained by the process described in IV from the previous amide with a yield of 40%. Its fusion point is  $180^{\circ}\text{C}$ . It is <sup>-in</sup>/soluble in water, soluble in alcohol.

B - Synthesis of p carboxy phenoxy alkyl carboxylic acids and their derivatives

I - Synthesis of the acids

The p carboxy phenoxy alkyl carboxylic acids are obtained in the following manner :

- 3 moles of pellets of soda, 1 mole of p hydroxy benzoic acid and 1 mole of chloro-acid are brought to reflux in 1.3 litres of water then immediately acidified to PH3. The di-acid precipitates, is isolated by filtration, washed in water then alcohol, and the product has a yield of about 80%.

Examples :

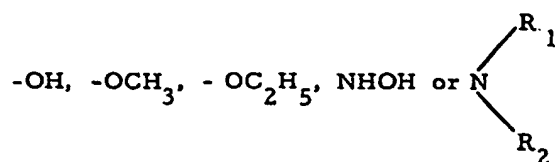
p carboxy phenoxy acetic acid is obtained by boiling to reflux with 120 g of soda, 138 g p hydroxy benzoic acid and 95g of chloroacetic acid in 1.3 litres of water, then acidification to PH3. After purification, 159 g of powder are obtained, its fusion point being  $282^{\circ}\text{C}$ .

II - Synthesis of ester, amide, and carbohydroxamic derivatives

It is possible to convert the series of p carboxy-phenoxy alkyl carboxylic acids into their ester, amide and carbohydroxamic derivatives according to the methods described above in AII, III, and VI. The compounds obtained conform to the general formula in such



a way that R can be



5  $\text{R}_1$  and  $\text{R}_2$  can be linear radicals such as  $\text{CH}_3$  or  $\text{CH}_2\text{CH}_3$  or represent a ring such as that of piperidine, methyl 2 piperidine, piperazine, morpholine, pyrrolidine methyl 4 piperidine, N-phenyl piperazine, N-p methoxy phenyl piperazine N-methyl 4 piperazine, N-p chlorophenyl piperazine, hexamethylemine or ethylamino ethylamine.

10 Y can also be -NHOH as indicated above in A VI

#### 1) Esterification

The di-acids obtained can be esterified according to the general method described in A II. The carboxyl in para position can also be esterified by maintaining the di-acid in suspension in the  
15 chosen alcohol at reflux under a current of HCl.

#### Examples :

a) p carboxy phenoxy ethyl acetate is obtained by boiling to reflux the previous body in a sulphuric medium in a mixture of ethyl alcohol and benzene. After distillation of the excess ethyl alcohol  
20 and solvent, the ester is isolated and recrystallised in benzene. The product is obtained with a yield of about 60%; its fusion point is  $138^\circ\text{C}$ ; it is soluble in alcohol and in bicarbonated water, and insoluble in water.

b) The ethylic di-ester of p carboxy phenoxy acetic acid is obtained by boiling to reflux a saturated solution of HCl containing 75 g of di-acid in 500 ml of ethyl alcohol under constant bubbling  
25 with HCl. The di-acid dissolves slowly, reflux and the current of HCl being continued 2 to 4 hours after completion of the dissolving.

The solution is discoloured or black, then evaporated in a vacuum, the oil obtained put into a solution with ether and washed in water, then with water saturated with sodium bicarbonate; after evaporation of the solvents the oil is distilled in a vacuum, its boiling point is  $125^{\circ}\text{C}$  at 05 mm Hg. The ester forms into a mass; 50 g of product is obtained and its fusion point is  $32^{\circ}\text{C}$ . It is soluble in alcohol and in ether, insoluble in water.

## 2 - Amidification

Amidification is obtained according to the general method described in A III.

### Examples :

a) (p carboxy phenoxy acetyl) 1 piperidine is obtained according to the method described in A III (1), from the ester described above in B II (1a) with a yield of about 45%. Its fusion point is  $190^{\circ}\text{C}$ ; it is soluble in alcohol and in bicarbonated water, insoluble in water.

b) (p methyl oxy carbonyl phenoxy acetyl) 1 piperidine is obtained from the methyl di-ester of carboxy phenoxy acetic acid, this being obtained according to the method described in BII (1). Amidification is carried out according to the general method described in A III; the yield is in the order of 74%. The product has a fusion point of  $104^{\circ}\text{C}$ . It is soluble in alcohol and ether, insoluble in water and petroleum ether.

c) (p acetyl oxy carboxyl phenoxy acetyl) 1 piperidine is obtained from the ethylic di-ester described above in B II (1b) with a yield of the order of 70%, and having a fusion point of  $61^{\circ}\text{C}$ . It is soluble in alcohol and ether, insoluble in water and petroleum ether.

d) (p carboxy phenoxy acetyl) 1 morpholine is obtained according to the general method of amidification from the ester obtained as above in B II (1a), with a yield of 55%. It is soluble in alcohol and bicarbonated water, insoluble in water; its fusion point is 183°C.

5 e) (p piperidino oxo) phenoxy ethyl acetate is obtained from the ester described in B II (1a). Previously chlorinated into p by the action of PCl 5 in equimolecular proportions under heat. The chloride of the acid obtained is made into a benzene solution, and a solution of piperidine in benzene is added to the ice bath in the proportions of  
10 1 : 2. The amide is separated by concentration in a vacuum, and recrystallisation. It is obtained with a yield of 40%. Its fusion point is 90°C. It is soluble in alcohol and in ether, insoluble in water.

f) p carboxamide phenoxy acetamide is obtained from carboxy phenoxy acetic acid, previously converted into its dichloro-  
15 acid derivative, this being put in the presence of NH<sub>4</sub> OH. The diamide precipitates; after purification the yield is 70%. Its fusion point is 265°C and it is insoluble in the usual solvents.

### 3). Hydroxamic acids

The corresponding hydroxamic acids are obtained  
20 according to the general method described in A VI

THE EMBODIMENTS OF THE INVENTION IN WHICH AN EXCLUSIVE  
PROPERTY OR PRIVILEGE IS CLAIMED ARE DEFINED AS FOLLOWS:

1. Process for preparing compounds having useful pharmaceutical properties of formula

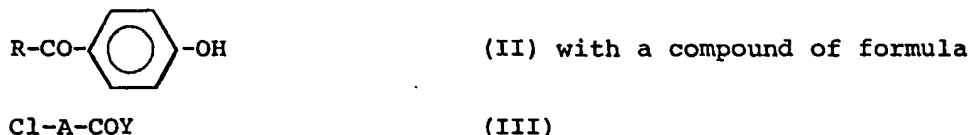


A is selected from the group consisting of  $\text{C}(\text{CH}_3)_2$  and a phenyl substituted alkyl chain having from 1 to 4 C;

R is selected from H, free or W-halogenated  $\text{CH}_3$ ,  $\text{C}_2\text{H}_5$  and  $\text{C}_3\text{H}_7$  groups and  $\text{C}_6\text{H}_5$ ; and

Y is selected from OH, lower alkoxy having 1 to 3 C, aliphatic amines selected from the group consisting of dimethyl-, diethyl-, ethylamino-ethylamine, and heterocyclic amines selected from the group consisting of piperidine, 2-methyl piperidine, 4-methyl piperidine, morpholine, pyrrolidine, piperazine, N-phenyl-, N-methyl, N-p-methoxyphenyl-, N-p-chlorophenyl-piperazine and hexamethyleneimine,

which comprises reacting a compound of formula



in the presence of a basic condensing agent, the compound (III) being selectively generated in situ when A is  $\text{C}(\text{CH}_3)_2$  from anhydrous  $\text{CHCl}_3$  and acetone, and compound (I) being optionally eventually converted into the corresponding oxime.

2. A process according to claim 1, wherein Y is OH, which comprises esterifying the acid by means of a lower alkanol YOH.

**B**

3. A process according to claim 1, wherein Y is a lower alkoxy, which comprises ~~reaching the ester~~ <sup>reacting ester</sup> with a secondary or heterocyclic amine HY to convert the ester into an amide.

4. A process according to claim 1, wherein compounds (I) are converted into oximes by means of hydroxylamine hydrochloride.

**A**

5. Pharmaceutical valuable compounds of the following general formula:

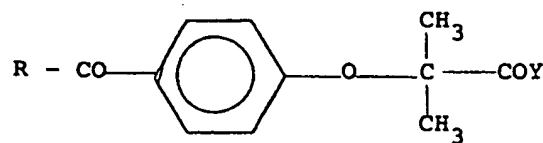


A is selected from the group consisting of  $\text{C}(\text{CH}_3)_2$  and a phenyl substituted alkyl chain having from 1 to 4 C;

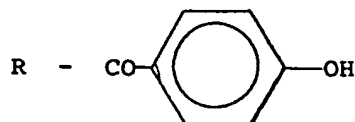
R is selected from H, free or W-halogenated  $\text{CH}_3$ ,  $\text{C}_2\text{H}_5$  and  $\text{C}_3\text{H}_7$  groups and  $\text{C}_6\text{H}_5$ ; and

Y is selected from OH, lower alkoxy having 1 to 3 C, aliphatic amines selected from the group consisting of dimethyl-, diethyl-, and ethylamino-ethylamine, and heterocyclic amines selected from the group consisting of piperidine, 2-methyl piperidine, 4-methyl piperidine, morpholine, pyrrolidine, piperazine, N-phenyl-, N-methyl, N-p-methoxyphenyl-, N-p-chlorophenyl-piperazine and hexamethylenimine, when prepared according to the process of claim 1 or its obvious chemical equivalent.

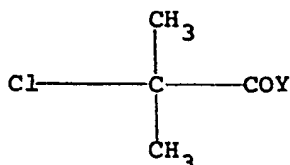
6. Process for preparing compounds having useful pharmaceutical properties of the general formula



wherein R is selected from the group consisting of H, CH<sub>3</sub>, C<sub>2</sub>H<sub>5</sub>, C<sub>3</sub>H<sub>7</sub> and C<sub>6</sub>H<sub>5</sub> and Y is selected from the group consisting of OH, OCH<sub>3</sub>, OC<sub>2</sub>H<sub>5</sub>, and OC<sub>3</sub>H<sub>7</sub> comprising reacting a compound of the formula

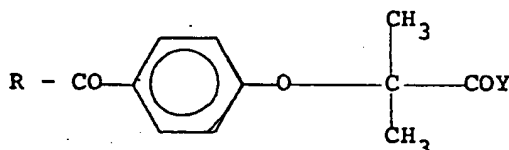


with a compound of the formula



wherein R and Y are as above defined, in the presence of a basic condensing agent.

7. Pharmaceutically valuable compounds of the general formula



wherein R and Y are as defined in claim 6, when prepared according to the process of claim 6 or its obvious chemical equivalent.

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